## S1: Subtraction of background signal of Ni-NTA layer from the observed spectra

Fig A shows raw SEIRA spectra, showing the main experiment and a control experiment without nanodiscs. The Fig 3 shown in the main text is a subtraction of both curves. Both spectra were obtained after initialization of transcription/translation (time $=0 \mathrm{~s}$ ) by adding the plasmid coding for bO.. The primary changes were positive peaks at $1564 \mathrm{~cm}^{-1}$ with a shoulder at $1590 \mathrm{~cm}^{-1}$ and 1417 $\mathrm{cm}^{-1}$ (Fig A(a)). Intensities of these peaks increased within ca. 30 seconds, decreased subsequently and vanished completely after 5 minutes. These bands are assigned to partial deprotonation of carboxylic groups (the asymmetric and symmetric $\mathrm{COO}^{-}$stretching modes, respectively) of $\mathrm{Ni}-$ NTA molecules underneath the nanodisc layer (Fig B). The addition of the provided feeding mix solution with the plasmid may differ in pH from the buffer solution ( pH 7.4 ) overlaying the Au surface, leading to a short pH jump visible in the spectra. As appearance of the bands from Ni-NTA layer hampered the analysis of protein, which appears in the same spectral regions we subtracted both spectra in the manuscript.


Figure A: A set of raw SEIRA spectra relevant to figure 3 in main text. Representative spectra obtained in each phase after initialization of transcription/translation by adding the plasmid encoding bO. Solid curves represent insertion and folding of the expressed proteins in the nanodisc monolayer. Broken curves represent results of a control experiment that has been done under the same conditions but without a nanodisc monolayer on the Au surface.


Figure B: pH induced SEIRA difference spectrum of Ni-NTA SAM layer. Reference spectrum was taken in 20 mM HEPES, 150 mM NaCl buffer with $\mathrm{pH}=8.5 .1 / 10$ volume of 0.1 M HCl was added to the cell,which led to a final $\mathrm{pH}=7.5$. This spectrum is only observed within the first 10 seconds after addition of HCl , when the local pH in the vicinity of the Au surface becomes momentarily lower.

